

## Sample Question Paper (Thermodynamics-CBCGS)

- In an open system, for maximum work, the process must be entirely
  - irreversible
  - reversible
  - adiabatic
  - none of the mentioned
- When a system changes its state from one equilibrium state to another equilibrium state, then the path of successive states through which the system has passed, is known as
  - Thermodynamic law
  - Thermodynamic process
  - Thermodynamic cycle
  - None of these
- Otto cycle consists of
  - Two constant volume and two isentropic processes
  - Two constant pressure and two isentropic processes
  - Two constant volume and two isothermal processes
  - One constant pressure, one constant volume and two isentropic processes
- The efficiency of a gas turbine is given by
  - (Net work output)/(Work-done by the turbine)
  - (Net work output)/(Heat supplied)
  - (Actual temperature drop)/(Isentropic temperature drop)
  - (Isentropic increase in temperature)/(Actual increase in temperature)
- Which of the following is true for a steady flow system?
  - mass entering = mass leaving
  - mass does not enter or leave the system
  - mass entering can be more or less than the mass leaving
  - none of the mentioned
- If the value of  $n = 0$  in the equation  $pv^n = C$ , then the process is called
  - Constant volume process
  - Adiabatic process
  - Constant pressure process
  - Isothermal process
- A piston cylinder contains air at 600 kPa, 290 K and a volume of  $0.01\text{m}^3$ . A constant pressure process gives 54 kJ of work out. Find the final volume of the air.
  - $0.05\text{ m}^3$
  - $0.01\text{ m}^3$
  - $0.10\text{ m}^3$
  - $0.15\text{ m}^3$
- A hot gas flowing through a pipeline can be considered as a
  - reversible process
  - irreversible process
  - both of the mentioned
  - none of the mentioned
- The latent heat of steam at atmospheric pressure is.....
  - 1535 kJ/kg

- (B) 1875 kJ/kg
  - (C) 2257 kJ/kg
  - (D) 2685 kJ/kg
10. Which of the following laws is applicable for the behaviour of perfect gas
- (A) Boyle's law
  - (B) Charle's law
  - (C) Gas-Lussac law
  - (D) All of the above
11. Otto cycle is a.....
- (A) Constant pressure cycle
  - (B) Constant volume cycle
  - (C) Constant teperature cycle
  - (D) Constant entropy cycle
12. For the same compression ratio,the efficiency of diesel cycle is.....otto cycle
- (A) Greater than
  - (B) Less than
  - (C) Equal to
  - (D) None of the above
13. The locus of standard liquid line and standard vapour line meets at.....
- (A) Boiling point
  - (B) Critical point
  - (C) Ice point
  - (D) Triple point
14. According to kinetic theory of gases,the absolute zero teperature is attained when.....
- (A) Volume of gas is zero
  - (B) Pressure of the gas is zero
  - (C) Kinetic energy of the molecules is zero
  - (D) Specific heat of gas is zero
15. The unit of pressure in S.I. unit is.....
- (A) Kg/cm<sup>2</sup>
  - (B) mm of water column
  - (C) Pascal
  - (D) Bars
16. An closed system is one in which.....
- (A) Mass does not cross boundaries of the system,through energy may do so
  - (B) Neither mass nor energy crosses the boundsries of the system
  - (C) Both energy and mass cross the boundaries of the system
  - (D) Mass crosses the boundary but not the energy
17. Specific heat of air at constant pressure is equal to.....
- (A) 0.17
  - (B) 0.21
  - (C) 0.24
  - (D) 1.0
18. Characteristic gas constant of a gas is equal to.....
- (A)  $C_p/C_v$
  - (B)  $C_v/C_p$
  - (C)  $C_p - C_v$
  - (D)  $C_p + C_v$

19. The unit of energy in S.I. unit.....
- (A) Watt
  - (B) Joule
  - (C) Joule/sec
  - (D) Joule/m
20. When cut-off ratio is \_\_\_\_\_ the efficiency of Diesel cycle approaches to Otto cycle efficiency.
- (A) Zero
  - (B) 1/5
  - (C) 4/5
  - (D) 1
21. The gas turbine cycle with regenerator improves
- (A) Thermal efficiency
  - (B) Work ratio
  - (C) Avoids pollution
  - (D) None of these
22. First law of thermodynamics deals with
- (A) Conservation of heat
  - (B) Conservation of momentum
  - (C) Conservation of mass
  - (D) Conservation of energy
23. Which of the following is a reversible non-flow process?
- (A) Isochoric process
  - (B) Isobaric process
  - (C) Hyperbolic process
  - (D) All of these
24. If in the equation  $pvn = C$ , the value of  $n = \infty$ , then the process is called
- (A) Constant volume process
  - (B) Adiabatic process
  - (C) Constant pressure process
  - (D) Isothermal process
25. The measurement of a thermodynamic property known as temperature is based on
- (A) First law of thermodynamics
  - (B) Second law of thermodynamics
  - (C) Zeroth law of thermodynamics
  - (D) None of these
26. The compression ratio is the ratio of
- (A) Total volume to swept volume
  - (B) Swept volume to clearance volume
  - (C) Swept volume to total volume

(D) Total volume to clearance volume

27. In a reversible isothermal expansion process, the fluid expands from 10 bar and 2 m<sup>3</sup> to 2 bar and 10 m<sup>3</sup>. During the process the heat supplied is at the rate of 100 kW. What is the rate of work done during the process?
- (a) 20 kW
  - (b) 35 kW
  - (c) 80 kW
  - (d) 100 kW
28. An ideal gas at 27°C is heated at constant pressure till its volume becomes three times. What would be then the temperature of gas?
- (a) 81° C
  - (b) 627° C
  - (c) 543° C
  - (d) 327° C
29. In a reversible isothermal expansion process, the fluid expands from 10 bar and 2 m<sup>3</sup> to 2 bar and 10m<sup>3</sup>, during the process the heat supplied is 100 kW. What is the work done during the process?
- (a) 33.3 kW
  - (b) 100 kW
  - (c) 80 kW
  - (d) 20 kW
30. The value of compressibility factor for an ideal gas may be: [IES-2002] 1. less or more than one 2. equal to one 3. zero 4. less than zero The correct value(s) is/are given by:
- (a) 1 and 2
  - (b) 1 and 4
  - (c) 2 only
  - (d) 1 only
31. Gauge pressure of air to which the ball must have been originally inflated so that it would equal 1 bar gauge at the stadium is:
- (a) 2.23 bar
  - (b) 1.94 bar
  - (c) 1.07 bar
  - (d) 1.00 bar
32. A 100 W electric bulb was switched on in a 2.5 m × 3 m × 3 m size thermally insulated room having a temperature of 20°C. The room temperature at the end of 24 hours will be
- (a) 321°C
  - (b) 341°C
  - (c) 450°C
  - (d) 470°C
33. A Carnot engine rejects 30% of absorbed that to a sink at 30° C. The temperature of the heat source is
- (a) 100° C

- (b) 433° C
- (c) 737° C
- (d) 1010° C

34. A reversible heat engine rejects 50 percent of the heat supplied during a cycle of operation. If this engine is reversed and operates as a heat pump, then what is its coefficient of performance?

- (a) 1.0
- (b) 1.5
- (c) 2.0
- (d) 2.5

35. A heat engine is supplied with 250 kJ/s of heat at a constant fixed temperature of 227°C; the heat is rejected at 27°C, the cycle is reversible, then what amount of heat is rejected?

- (a) 250 kJ/s
- (b) 200 kJ/s
- (c) 180 kJ/s
- (d) 150 kJ/s